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TITLE

GAS LAW PROGRAMS

AUTHOR

Thomas Ford

COMPANY

White Mountains Regional High School Whitefield, New Hampshire

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SOURCELANGUAGE

FOCAL

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GAS LAW PROGRAMS

DECUS Program Library Write-up

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FOCAL -

GAS LAW PROGRAMS

AUTHOR: Thomas J. Ford

White Mountains Regional High School Whitefield, N. H. 03598

CONSISTING OF:

IDEAL GAS PLOT P/V

IDEAL GAS VOLUME VS TEMPERATURE

REAL GAS VOLUME VS TEMPERATURE

IDEAL GAS PLOT P/V

ABSTRACT:

The experimental plot of pressure vs volume for any real gas sample yields data which do not fall on a straight line. Since there are typically too few data to get a good idea of the indicated curve, some assistance (or insight) is required.

This FOCAL program plots the relationship of an inverse proportionality and simultaneously plots the reciprocal of this relationship. Depending upon the input parameters, plotting can be varied to examine different portions of the inverse proportionality curve. In each instance, the reciprocal of this relationship plots as a stright line, whose slope can be varied to facilitate display and interpretation. The program will label the axes, whether or not the data start from zero on the pressure scale. Since group 1 serves no useful purpose other than to convey a suscinct set of rules for using the program, it can be erased at any time to facilitate reentry into the operational program.

There has been no attempt made to "rig" the program to realistically handle experimental data since the intent was to be able to play with the input parameters intuitively in the search for insight and understanding.

The obvious experimental conclusion would be to plot pressure vs the reciprocal of the experimental volume. If this were to yield a straight line (which is easier to determine since there is only one kind) then the original data must have been inversely proportional, and BOYLE'S LAW is corroborated. The manipulation is necessary since there are at least a large number of curves which might be argued from the limited data.

```
01.01 T "IDEAL GAS IN QUADRANT ONE",!
01.02 T "PLOT OF PRESSURE VS VOLUME",!
01.03 T "GIVE POSITIVE VALUES FOR:",!
01.04 T " LOWEST PRESSURE"!" INCREMENT"!" HI PRESSURE"!
01.05 T " K IN PV=K",!," SLOPE OF RECIPROCAL PLOT",!
01.06 C ARBITRARY VALUES AND UNITS USED FOR PARAMETERS
01.10 C ERASE 1 LEAVES ESSENTIALS OF PROGRAM
02.02 A ?LPRESS, IPRESS, UPRESS, K, SLOPE?,!
02.04 I (LP)1.03,6.01,2.06
02.06 DO 8;F P=2, IP, LP;T ":"!
06.01 F P=LP, IP, UP; DO 7
07.01 I (P)7.02,7.15,7.02
07.02 T ":",#
07.05 F V=0,(K/P);T " "
07.08 T "*",#
07.11 F V=0, (P/K)*SLOPE; T " "
07.14 T "O",!;R
07.15 DO 8;R
08.01 T " ",#
08.02 T !"P"!"R"!"E"!"S"!"S"!"U"!"R"!"E"!!"A"!"X"!"I"!"S"!!
08.03 T "===VOLUME AXIS======DATUM OUT THERE SOMEWHERE====(*)"!;R
```

```
*6
IDEAL GAS IN QUADRANT ONE
PLOT OF PRESSURE VS VOLUME
GIVE POSITIVE VALUES FOR:
LOWEST PRESSURE
INCREMENT
HI PRESSURE
  IN PV=K
SLOPE OF RECIPROCAL PLOT
LPRESS,:10 IPRESS,:1 UPRESS,:100 K,:700 SLOPE:400
      0
        0
        O
        . 0
:
          0
          0
           U
           O
            0
            U
             ()
              0
              O
               O
               U
                O
                0
                  O
                   O
                   ()
                    0
                    Ö .
                     ()
                      0*
                       0
                      *()
                        O
                        0
                          0
                          O
                           0
                            0
                            0
                             0
                             0
                              ()
                              0
                               ()
                                0
                                0
                                 0
                                 0
                                   0
                                   0
                                    O
                                     ()
                                     (1
```

```
*GO
 IDEAL GAS IN QUADRANT ONE
 PLOT OF PRESSURE VS VOLUME
 GIVE POSITIVE VALUES FOR:
     LOWEST PRESSURE
     INCREMENT
     HI PRESSURE
     K IN PV=K
     SLOPE OF RECIPROCAL PLOT
LPRESS,: Ø IPRESS,: 1 UPRESS,: 2Ø K,: 7Ø SLOPE:
                                                                  1øø
P
R
E
S
S
U
R
E
A
X
I
===VOLUME AXIS=======DATUM OUT THERE SOMEWHERE====(*)
   0
 COMMENT * IS PLOT OF P VS V
 COMMENT O IS PLOT OF 1/P VS V
 COMMENT
          IF THE O'S PLOT A STRAIGHT LINE, THE *'S DESCRIBE
          AN INVERSE PROPORTIONALITY.
```

```
*GO
IDEAL GAS IN QUADRANT ONE
PLOT OF PRESSURE VS VOLUME
GIVE POSITIVE VALUES FOR:
     LOWEST PRESSURE
     INCREMENT
     HI PRESSURE
     K IN PV=K
     SLOPE OF RECIPROCAL PLOT
LPRESS,:Ø IPRESS,:1
                     UPRESS,:3Ø
                                    K,:7Ø
                                              SLOPE:7Ø
P
R
E
S
U
R
E
A
X
I
S
===VOLUME AXIS=======DATUM OUT THERE SOMEWHERE====(*)
```

```
G
LPRESS,:1Ø IPRESS,:35
*G
                          UPRESS,: ?Ø1.ØØ @ Ø2.Ø2
LPRESS,:1Ø IPRESS,:1
                           UPRESS,:35
                                           K,:5ØØ SLOPE:9Ø
P
R
E
s
s
U
R
E
A
X
Ι
S
===VOLUME AXIS=======DATUM OUT THERE SOMEWHERE====(*)
   0
    0
    0000
    0 0 0 0
     0
     000
     0 0 0 0 0 0
```

```
G
LPRESS,:5 IPRESS,:1 UPRESS,:4 K,:6pp
                                            SLOPE: 3ØØ
P R E S S U R E
A
X
I
S
===VOLUME AXIS=======DATUM OUT THERE SOMEWHERE====(*)
 8
```

IDEAL GAS VOLUME VS TEMPERATURE

This FOCAL program explores the CHARLES LAW relationship for an ideal gas. On the presumption that the gasw will change by 1/273.16 of its volume at 0°C. for each 1°C. change in temperature, a gas volume is input and the temperature increment by which the sample will be reduced in temperature. The program will plot volume and temperature until it senses a negative volume. Since negative volume has no physical meaning, the temperature associated with it probably doesn't either, and the last temperature must be within the temperature increment of an absolute zero temperature. Provision is made to input a low temperature limit for the plot to add an element of discovery to the proceedings by enabling a realistic search for the zero volume intercept.

The initial volume is scaled to fit the TTY page, but the data table printed on each line gives the actual values of temperature and volume. Overprinting the data table is intentional to give maximum plotting space since the search for the intercept within 0.1°C. will yield a display about 12 yards long (and require about 4 hours on the PDP-8/S).

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- 01.02 T "IDEAL GAS VULUME VS TEMPERATURE"! 01.14 T "TEMP REDUCED STARTING AT " DEG. C."! 91.06 T "GIVE POSÍTIVE VALUES FOR:"! 91.08 T " - VOLUME AT & C. IN ARBITRARY UNITS"! 91 • 10 T " NEG. TEMP. INCREMENT"! 01.12 T " LOU TEMP. LIMIT OF PLOT"! 91.14 C ERASE 1 WILL RETAIN ESSENTIAL PROGRAM 02.02 S T=0; A "INITIAL VOL", IV, "TEMP INCR", IT, "LOW LIMIT", LT 02.03 T ""; C BELL LINE 02.04 T !"T"!"E"!"M"!"P"!!"A"!"X"!"I"!"S"!;S @V=IV;DO 6.09 02.06 S V=68; F M=0, IT, LT; DU 6 M2.87 I (OV)2.88,2.38;T "THIS IS NOT WITHIN TEMP INCR OF ABS ZERO";0 92.08 I "SINCE NEG VOLUME HAS NO PHYSICAL MEANING, TEMPERATURES"! 42.09 I "RELATED TO THESE WOULD HAVE NO PHYSICAL MEANING, AND OUR"!

02.19 T "ZERO VOLUME INTERCEPT MUST BE WITHIN THE TEMP INCREMENT"!

02.11 T "OF AN ABSOLUTE ZERO TEMPERATURE"!;0

FOCAL -

REAL GAS VOLUME VS TEMPERATURE

This FOCAL program accepts experimental sets of volume and temperature information, performs a LEAST SQUARES REGRESSION FIT to a straight line, and extrpolates and plots this line to a zero volume intercept. The temperature associated with the last positive volume as temperature is reduced is compared with the accepted value for absolute zero, and the percentage deviation typed out. The least squares regression program is also listed separately.

The magnitude of the extrapolation indicates that high percentages of deviation are to be expected from limited input data.

- 01.02 T "REAL GAS VOLUME VS TEMPERATURE"! 01-05 T "TEMP REDUCED STARTING AT DATA AVERG. TEMP"! NO. OF SETS OF 'T' AND 'V' DATA"! 01.08 T "GIVE 01 • 11 · T " EACH DATUM TEMP C."!. 31 - 14 1 " EACH DATUM VOLUME IN ARBITRARY UNITS"! 01 - 17 T " TEMP INCR FOR PLOT AND TABLE"! Ø1.18 T "'GO' AGAIN"!!;E 1 82.81 ["'CHARLES LAW' PLOT OF EXP. DATA TO DETERM ABS. ZERO"! 02.02 S SX=0; S SY=0; S SZ=0; S SM=0 02.05 A "NUMBER OF DATUM POINTS", No "PLOT INCR", IT,! 02.08 F K=1.N;A "T",X,"V",Y,!;D 3.02 02.11 6 3.05 03.02 S SX=SX+X;S SY=SY+Y;S SZ=SZ+(X*Y);S SM=SM+X+2 03.05 S M=[N*SZ-SX*SY]/[N*SM-SX+2]; S B=[SY-M*SX]/N 03.20 ["M", M," B",B,!" EXP. PLOT IS: V",M,"*T+",B,!!! 04.02 S T=SX/N; S OV=SY/N; S MP=70/(T+<B/M>) 04.05 T ""; C BELL LINE 04.08 T !"T"!"E"!"M"!"P"!!"A"!"X"!"I"!"S"!;DO 6.9;S V=68 $04 \cdot 11 \text{ F } Z = 0 \cdot [T + \langle B / M \rangle] / [T \cdot D] 6$ 04.19 T "SINCE NEG VOLUMES HAVE NO PHYSICAL MEANING, "!;6 4.2 04.20 T "NEITHER WOULD TEMPERATURES RELATED TO THEM"!; G 4.21 04.21 T "AND OUR ZERO VOLUME INTERCEPT MUST BE "!, % 4.01; G 4.22 04.22 T "WITHIN THE PLOT INCR OF AN ABSOLUTE ZERO "; G 4.23 24.23 T "TEMPERATURE"!"WHICH I CALCULATE TO BE: ", -B/M;D 7;0 06.08 T ":", %5.02 T," C. V ", OV, # 06.11 F K=0, V; T " " 06.14 T "(X)"!; S V=V-[IT*MP]; S QV=QV-[IT*M] 06.17 S T=T-IT;R 06.90 T ": I WILL PLOT FIRST VOLUME AT FULL SCALE"!

*

C-FOCAL , 8/68

- 01.10 T "CURVE FITTING BY LEAST SQUARE REGRESSION",!!
- 01.20 T "N=NUMBER OF DATUM POINTS",!
- 01.30 T "X=ABSCISSA, Y=ORDINATE",!
- 03.10 S SX=0;5 SY=0;S SM=0;S SE=0;S A1=0;S A0=0
- 03.20 A ?N?,!;F K=1,N;A ?X,Y?,!;D 6
- 04.10 S A1=[N*SM-SX*SY]/[N*SE-SX+2]
- 04.20 S A0=[SY-A1*SX]/N
- 05.10 T "A1",A1," A0",A0,!!!,"
- Y",A1,"*X+",A0,!;0

- 76.10 S SX=SX+X
- 06.20 S SY=SY+Y
- .06.30 S SM=SM+X*Y
- 06.40 S SE=SE+X+2
- * ବର୍ଷ ବର୍ଷ ପ୍ରତ୍ୟକ୍ତ ବର୍ଷ ଓ ଅନ୍ତର୍ଶ ପ୍ରତ୍ୟ ପ୍ରତ୍ୟ କ୍ଷ୍ୟ ହେଉ ଓ ଅନ୍ତର୍ଶ ବର୍ଷ ବର୍ଷ ବର୍ଷ ବର୍ଷ ବର୍ଷ ଓ ଅନ୍ତର୍ଶ ବର୍ଷ